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greater than 1.7 and is
a transfer of electrons.

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The Lower

Electronegativity

transfers to the higher
(electronegativity) t.

Ionic Bond. Ions pack in
a crystal lattice

structure. Lattice

Energy. Cation {+}

and anion {-} create

_____.

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Covalent & Molecular

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CHAPTER 6 REVIEW

Chemical Bonding

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ANSWER Answer the
following questions in
the space provided. 1.

b In metals, the
valence electrons are
considered to be (a)
attached to particular
positive ions. (c)
immobile.

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Name Section Quiz,
continued Class Date

6. Metals are malleable
because when struck,
one plane of metal
atoms can slide past
another plane without
breaking bonds. b.

cannot easily move out
of the way. c. moves in
a way that maximizes
the repulsive forces
within the

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Chemistry Chapter 6-
Chemical Bonding 1

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Chapter 6- Chemical
Bonding. chemical

bond- a mutual

electrical attraction

between the nuclei and
the valence electrons

of different atoms that
binds the atoms

together ; 2. ionic

bond- a chemical bond

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that results from the electrical attraction between anions and cations ...

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Class Date Chapter

Test B, continued 18.

When a strong acid is

titrated with a weak

base, the pH of the

solution at the

equivalence point is

than 7. 19. A is a highly

purified solid used to

check the

concentration of a

standard solution. 20.

A 1 M solution of NaOH

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will have a pH that is

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Bonding MIXED

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ANSWER Answer the
following questions in
the space provided. 1.

a.

6 Chemical Bonding - Effingham County Schools / Overview

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Structure #2
Simulations:
Applications: Blood
Chemistry
(Hemoglobin, Iron Use
and Storage, Dialysis in
Kidneys, pH regulation
during exercise),
Nutrients and
Solubility, Enzyme
Kinetics and Inhibitors
in HIV Drugs, Enzyme-
Substrate Binding,

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Vision and Light
Induced Molecular
Changes ...

Chemistry I - Mr. Benjamin's Classroom

Answer to 12. Each period in the periodic table corresponds to .
a. an energy level c. an orbital
b. an electron d. a suborbital

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