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Elements and Isotopes
A chemical element is

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a pure substance that consists entirely of one type of atom.

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Chapter 2: Introduction to the Chemistry of Life

Figure 2.1 Foods such as bread, fruit, and cheese are rich sources of biological

macromolecules. The elements carbon, hydrogen, nitrogen, oxygen, sulfur, and

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phosphorus are the key building blocks of the chemicals found in living things. <https://opentextbc.ca/biology/chapter/chapter-2-introduction-to-the-chemistry-of-life/> read more.

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chapter_2.1 - Chapter
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Answer Key 2.1 The

Nature of Matter

Lesson Summary

Atoms The atom is the basic unit of matter, made up of three subatomic particles.

Protons have a positive charge and neutrons carry no charge.

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CONCEPTS BIOLOGY

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Chemistry of Life 2.1

Atoms, Ions, and

Molecules All living

things are based on

atoms and their

interactions, 2.2

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Properties of Water

Water's unique

properties allow life to
exist on Earth. 2.3

Carbon-Based

Molecules Carbon-

based molecules are
the foundation of life.

2.4 ...

Chapter 2 The Chemistry Of Life Study Guide Answer Key

Chapter 2 The

Chemistry of Life

Section Review 2-1

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Section Review 2-4 1.
protons, neutrons 2.
electrons 3. neutrons
4. electrons 5. ionic 6.
The two main types of
chemical bonds are
ionic and covalent
bonds. https://mafiadoc.com/chapter-2-the-chemistry-of-life-answer-key_5a2ec8921723dd9433478200.html read
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Comment. 10. The
prop- erty of
radioactive isotopes
that is useful for dating
is that they break down
at a constant rate over
time. Section Review
2-2. 1. a 2. b 3. c 4. c
5. a 6.

Chapter 2 The
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Section Review 2-4 1.
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electrons 3. neutrons

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4. electrons 5. ionic 6.
The two main types of
...

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Chapter 22-23

Structural Formulas of
Alkanes SG 22.1 &

22.2 Structures of

Alkenes & Alkynes SG

22.3 & 22.4 SG 23.1,

23.2, 23.3 Chapter 22

Supplemental Problems

Structures of

Functional Groups

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Chapter 22 Review
Assessment-----

Chapter 3 SG 3.1 SG
3.2 SG 3.4 Chapter 3
Supplemental Problems
Chapter 3 Review
Physical and Chemical
Changes ...

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Chapter Vocabulary
Review 1. nucleus 2.

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element 3. compound
4. ionic, covalent 5.
van der Waals forces 6.

A sodium atom is a neutral particle that contains equal numbers of protons and electrons. A positive sodium ion is a sodium atom that has a positive charge because it has lost an ...

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Unit 1 The Chemistry of Life: Chapter 2 - Ms. Kots Biology ...

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as bread, fruit, and
cheese are rich sources
of biological
macromolecules. The
elements carbon,
hydrogen, nitrogen,
oxygen, sulfur, and
phosphorus are the key
building blocks of the

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chemicals found in
living things.

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The following Chapters
are included (based on
Essentials of Human
Anatomy & Physiology
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Marieb):Chapter 1 -
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OrientationChapter 2 -
Basic
ChemistryChapter 3 -
Cells and

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Section Review 2-4 1.

protons; neutrons 2.

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4. electrons 5. ionic 6.

The two main types of
chemical bonds are

ionic and covalent

bonds. Chapter 2 The

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your answers to

questions 19 through

21 on the particle

diagrams below, which

show atoms and/ or

molecules in three

different samples of

matter at STP 19

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Unit 2 Stoichiometry

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Example: How many

moles of salt are 7

formed from

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set (18) Scientific method, logical approach to solving problems by observing and collecting data, formulating hypothesis, testing hypothesis, forming theories that supported data.

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